

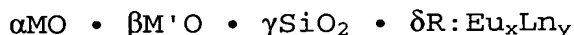
Amendments of the Claims

This listing of claims will replace all prior versions and listings of claims in the application.

Listing of Claims:

1. (currently amended) Light-storage self-luminescent glass, comprising from 0.01% to 40% by weight of a light-storage self-luminescent material activated by multiple ions and from 99.99% to 60% by weight of a matrix glass; wherein the light-storage self-luminescent material has a particle size from ~~0.55~~ 0.8 mm to 20 mm, and the matrix glass is selected from the group consisting of ~~common silicate~~ sodium-calcium-silicon glass, borate glass, phosphate glass, halide glass, sulfide glass and aluminate glass.

2. (original) Light-storage self-luminescent glass according to claim 1, wherein the chemical formula of the light-storage self-luminescent material activated by multiple ions is:



wherein M is one or more selected from the group consisting of Sr, Ca, Ba and Zn;

M' is one or more selected from the group consisting of Mg, Cd and Be;

R is B_2O_3 , P_2O_5 or mixture thereof;

Ln is one or more selected from the group consisting of Nd, Dy, Ho, Tm, La, Pr, Tb, Ce, Er, Mn, Bi, Sn and Sb; and

α , β , γ , δ , x and y are molar coefficients meeting following requirement: $0.6 \leq \alpha \leq 6$; $0 \leq \beta \leq 5$; $1 \leq \gamma \leq 9$; $0 \leq \delta \leq 0.7$; $0.00001 \leq x \leq 0.2$; $0 \leq y \leq 0.3$.

3. (currently amended) Light-storage self-luminescent glass according to claim 2, wherein the ~~main~~

chemical formula of the light-storage self-luminescent material activated by multiple ions is:

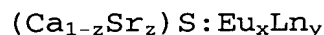


wherein Ln is one or more selected from the group consisting of La, Ce, Dy, Tm, Ho, Nd, Er, Sb and Bi;

z is a coefficient: $0 \leq z \leq 1$; and

x and y are molar coefficients: $0.0001 \leq x \leq 0.2$; $0.0001 \leq y \leq 3.0$.

4. (original) Light-storage self-luminescent glass according to claim 1, wherein the chemical formula of the light-storage self-luminescent material activated by multiple ions is:

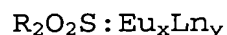


wherein Ln is one or more selected from the group consisting of Er, Dy, La, Tm and Y;

z is a coefficient: $0 \leq z \leq 1$; and

x and y are molar coefficients meeting following requirement: $0.00001 \leq x \leq 0.2$; $0.00001 \leq y \leq 0.15$.

5. (original) Light-storage self-luminescent glass according to claim 1, wherein the chemical formula of the light-storage self-luminescent material activated by multiple ions is:



wherein R is one or more selected from the group consisting of Y, La and Gd;

Ln is one or more selected from the group consisting of Er, Cr, Bi, Dy, Tm, Ti, Mg, Sr, Ca, Ba and Mn; and

x and y are molar coefficients meeting

following requirement: $0.00001 \leq x \leq 0.2$; $0.00001 \leq y \leq 0.6$.

6. (original) Light-storage self-luminescent glass according to claim 1, wherein the chemical formula of the light-storage self-luminescent material activated by multiple ions is:

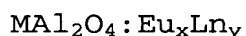


wherein M is one or more selected from the group consisting of Mg, Ca, Sr and Zn;

Ln is one or more selected from the group consisting of Nd, Dy, Ho, Tm, La, Ce, Er, Pr and Bi; and

α , β , γ , x and y are molar coefficients meeting following requirement: $0.5 \leq \alpha \leq 6$; $0.5 \leq \beta \leq 9$; $0 \leq \gamma \leq 0.3$; $0.00001 \leq x \leq 0.15$; $0.00001 \leq y \leq 0.2$.

7. (original) Light-storage self-luminescent glass according to claim 6, the chemical formula of the light-storage self-luminescent material is:

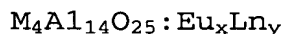


wherein Ln is one or more selected from the group consisting of La, Ce, Dy, Ho, Nd and Er;

M is one or more selected from the group consisting of Sr, Ca, Mg and Zn; and

x and y are molar coefficients: $0.0001 \leq x \leq 0.15$; $0.0001 \leq y \leq 0.2$.

8. (original) Light-storage self-luminescent glass according to claim 6, wherein the chemical formula of the light-storage self-luminescent material activated by multiple ions is:



wherein Ln is one or more selected from

the group consisting of Pr, Ce, Dy, Ho, Nd and Er;

M is one or more selected from the group consisting of Sr, Ca, Mg and Zn; and

x and y are molar coefficients: $0.0001 \leq x \leq 0.15$; $0.0001 \leq y \leq 0.2$.

9. (canceled)

10. (currently amended) Light-storage self-luminescent glass according claim 1, wherein the ~~common silicate~~ matrix glass consists of (by weight):

SiO ₂ : 30-81%	CaO: 0.5-9%
Al ₂ O ₃ : 0-23%	MgO: 1-8%
B ₂ O ₃ : 0-15%	SrO: 1-10%
Li ₂ O: 0-8%	BaO: 0-16%
Na ₂ O: 0.6-18%	ZnO: 0.6-55%
K ₂ O: 0.4-16%	PbO: 0-33%
As ₂ O ₃ : 0-0.5%.	

11. (canceled)

12. (currently amended) A process for producing the light-storage self-luminescent glass according to claim 1, comprising:

heating and melting the matrix glass;
doping the light-storage self-luminescent material into the melted matrix glass to produce a mixture; and
forming the mixture at 900-1300°C.

13. (previously presented) A process for producing the light-storage self-luminescent glass according to claim 1, comprising:

re-heating and melting a glass which has been formed and cooled; and

doping the glass with the light-storage self-luminescent material before secondary forming.

14-15. (canceled)

16. (new) Light-storage self-luminescent glass according claim 1, wherein said light-storage self luminescent material activated by multiple ions is selected from the group consisting essentially of silicate, aluminate, sulfide, and any combination thereof.